THE EFFECT OF MOISTURE ON THE PHYSICAL CHARACTERISTICS OF RANITIDINE HYDROCHLORIDE TABLETS PREPARED BY DIFFERENT BINDERS AND TECHNIQUES

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ABSTRACT

The effect of moisture on the physical properties of ranitidine hydrochloride tablets prepared by direct-compression and by wet-granulation using or EC as binders was studied. Tablets adsorped moisture at 50 and 75 % RH (relative humidity) but lost moisture at 30% RH. Except storage at 75% however, tablet volumes did not change significantly

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during the test period. Moisture sorption caused decrease in strength of tablets except low humidity (30% RH). Also, the disintegration time of showed a decrease at all conditions except Furthermore, generally dissolution profiles of prepared by direct-compression and by ethyl cellulose remained unchanged. Changes in the binder type tablet formulations changed the water uptake properties also and physical the properties of Directly-compressed tablets were much susceptible change caused by humidity than tablets prepared by wet-granulation.

INTRODUCTION

Ranitidine hydrochloride is widely used in treatment of peptic ulcers related disorders. and current literature does not contain reference that deal with the physical stability rantidine tablets. Previously, the effect of the on physical characteristics of ranitidine hydrochloride in powder and granule forms was reported (1).

On the other hand, earlier studies on the effect of moisture on tablet properties indicated wide response differences among the formulation and tabletting methods (2-4).

The present study was undertaken to investigate the effect of moisture sorption and desorption on the physical properties of ranitidine tablets. The changes in tablet characteristics upon aging in correlation to tabletting methods and binder type were also assessed.



Table 1 Tablet Formulations

Turnediant	Fo	rmulation	n (mg)
Ingredient	F-I	F-II	F-III
Ranitidine hydrochloride	150	150	150
Avicel PH 101	75	75	75
Anhydrous Lactose	25	25	25
Corn starch	48	48	48
Polivinyl pyrrolidone*		q.s.	
Ethyl cellulose**			q.s.
Magnesium stearate:Talc (1:9)	2.9	16.8	16.8

- Alcoholic solution of PVP (%5)
- ** Alcoholic solution of EC (%5)

EXPERIMENTAL

Materials

hydrochloride (Deva, Turkey) Ranitidine anhydrous lactose (Sheffield Chem. USA), Avicel (FMC, USA), polyvinylpyrrolidone (PVP K-30, BASF, FRG), ethyl cellulose (EC) (Prodotti-fromi, Italy).

Methods

Tablets were prepared by direct compression and wet-granulation methods. Formulations were given Table 1.



For wet-granulation, after mixing the powders, the granulating solution was added while mixing ovendried. The dry granulation was screened with magnesium stearate-talc mixtures.

Tablets were compressed on a machine (Manesty D-33, UK) using 10.4 mm diameter faced punches.

Tablets were stored on open petri dishes at 30, 50, 75 and 90% RH chambers and also RT/Ambient RH.

The samples were removed at regular intervals and evaluated over 120 days for their moisture content, diameter and thickness, crushing disintegration time and dissolution characteristics.

Moisture contents during the test determined as the weight gained and lost were as per cent water in the tablets. crushing strength of tablets were determined TBH-28, FRG) and the mean value of 6 tablets reported. Disintegration time of tablets was measured by the USP XXI method. Dissolution test was according to basket method of USP XXI. The drug content was determined by spectrophotometrically Techtron 634).

RESULTS and DISCUSSION

Moisture Content

Moisture adsorption levels of hydrochloride tablets were given in Table 2. At 90% RH,



Table 2 Moisture Levels (%)* of Ranitidine hydrochloride Tablets after Storage different Relative Humidities

T:						Formu:	lation								
Time (days)	 	F-	-1			F-	II	·		F-:	III				
	30%	50%	75%	90%	30%	50%	75%	90%	30%	50%	75%	90%			
í	-0.196	0.112	0.950	10.52	-0.027	0.585	2.070	9.45	-0.227	0.322	1.542	5.61			
2	-0.229	0.159	1.284	18.83	-0.040	0.694 2.460 17.89 -0.226 0.392 1	1.907	07 12.00							
5	-0.270	0.189	1.275	-	-0.018	0.703	3.070	0700.283 0.412 2	2.306)6 -					
8	-0.178	0.181	1.237	-	-0.069	0.686	3.690	-	-0.247	0.402	2.248	-			
14	-0.218	0.163	1.223	-	-0.105	0.641	3.930	-	-0.277	0.372	2.290	-			
30	-0.221	0.166	1.299	-	-0.142	0.649	4,450	-	-0.280 0.375	2.318 -	-,				
45	-0.224	0.164	1.304	-	-0.151	0.151 0.657 4.440) -	-0.299 0.361	2.346	-					
60	-0.230	0.169	1.309		-0.148	0.648	4.450 - 4.450 -	4.450 -	0.30	-0.305	05 0.339	2.341	-		
90	-0.225	0.167	1.301	-	-0.144	0.651		0 -	0.310	0.347 2.345	2.345	-			
120	-0.228	0.166	1.296	-	-0.146	0.656	4.441	0.312 0.350 2.3				0 -			

The mean of 2 values.

during the first 24 hrs, all the tablets rapidly water adsorbed and softened. As seen in the table, at 50 and 75% RH, tablets prepared by direct compression adsorbed less water than tablets made by wet-granulation. contrast, all tablets lost the water at 30% RH, however the less water desorption was observed in prepared with PVP. In general, except storage RH, direct-compressed tablets showed the lowest sorption. Among the studied binders, tablets made with



PVP adsorbed more moisture than those made This effect was attributed to the hygroscopic nature of PVP.

depended The amount of moisture adsorbed upon the type of binder used in ranitidine tablets. This data confirms the earlier report (5).

Tablet Volume

Except exposure to 75% RH, tablet volumes did not alter during the storage period (Table 3). RH, an extensive volume expansion was observed formulations. In particular, tablets prepared direct compression indicated the highest increase the other volume as seen in the table. On directcompressed tablets indicated the lowest uptake but the highest volume increase. This may be due to the effect of binders in wet-granulation methods.

Crushing Strength

As seen in Fig.1, the crushing strength tablets decreased when they were stored the decrease However, the extent of was wet-granulated tablets in accordance with their moisture uptake. It has been shown that free exists in solids in 2 states; a "pendular" state liquid bridges occur between indivindual particles a "capillary" state where all the pores of are filled with liquid. The significant decrease strength could be explained by presence of capillary water, though which interparticulate bonds are by dissolution (6). Therefore, reduction in strength of ranitidine hydrochloride tablets can be attributed



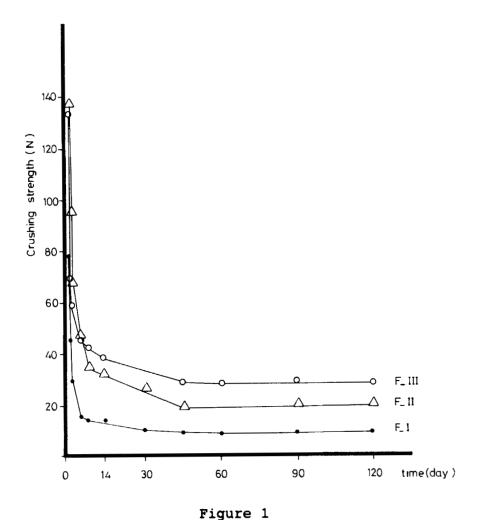
Table 3 Effect of Moisture on Tablet Volumes (mm3) (n=10)

Time					% RH	ک ایپا ہے ہوائی باپ ہے م		**********	
(days)		F-1			F-11			F-111	
	30	50	75	30	50	75	30	50	75
0	332.7	330.4	343.1	319.5	319.7	328.0	315.7	316.3	317.7
1	329.8	329.5	342.0	319.5	321.5	334.7	316.9	317.7	323.0
2	331.3	330.4	356.7	319.5	323.3	334.7	316.9	317.1	324.5
5	331.9	333.6	385.7	321.5	326.2	341.3	319.2	320.7	332.6
8	330.6	332.7	380.8	319.5	322.6	339.2	316.3	317.1	329.1
14	332.1	332.1	399.0	320.1	322.6	342.2	317.1	318.0	332.0
30	329.8	331.9	400.3	320.3	321.2	349.0	315.7	318.0	333.5
45	330.6	332.7	402.1	320.9	322.0	358.1	315.7	319.5	337.8
60	332.1	332.1	408.3	319.5	323.5	365.2	316.5	318.6	339.9
90	331.3	331.3	409.4	318.8	321.8	349.2	315.9	317.1	336.5
120	332.1	332.1	406.2	320.3	322.0	355.9	315.9	317.7	335.6

the capillary effect of moisture. Furthermore, decrease in strength may also be due to the significant of the compacts which increase in moisture content would separate the crystals from each other and thereby minimize the effect of the inherent forces and cohesion as noted by Lordi and Shiromani (7).

Table 4 summarized the crushing strength of tablets stored at 50% RH. At this condition,





Strength of Ranitidine Changes in Crushing hydrochloride Tablets at 75 % RH Key: F-I (•), F-II (Δ); F-III (0)

observed with considerable reduction was direct-compressed tablet whereas a slight decrease was seen in tablets made with EC. However the strength tablets made with PVP did not change.

Except direct-compressed tablets, there no significant change in tablet strength after storage 30% RH (Table 4).



Table 4 Effect of Moisture on the Crushing Stength $(N \pm SD)$ of Tablets

Tipe					% RH				
(days)		F-I			F-II	******		F-III	
	30	50	Amb	30	50	Amb	30	50	Amb
0	91.16 (6.958)	91.16 (6.958)	91.16 (6.958)	138.33 (7.637)	138.33 (7.637)	138.33 (7.637)	134.16 (9.242)	134.16 (8.465)	134.16 (9.242
1	79.75 (9.504)	74.83 (9.827)	-	127,25 (4,072)	118.87 (5.807)	-	116.25 (7.500)	112.00 (3.559)	-
2	72.50 (8.779)	62.25 (7.071)	-	138.66 (3.535)	115.62 (3.145)	-	118.00 (9.848)	106.62 (5.121)	-
8	71.75 (2.516)	85.70 (11.750)	-	137.83 (8.838)	128.70 (10.986)	-	131.50 (4.272)	113.20 (2.252)	-
14	82.00 (5.338)	81.75 (2.901)	-	135.87 (5.452)	123.25 (7.836)	-	130.25 (1.842)	116.12 (6.383)	-
30	86.00 (6.940)	71.90 (6.618)	65.30 (3.013)	137.80 (4.500)	124.50 (7.234)	125.70 (5.345)	139.00 (7.675)	118.10 (3.629)	115.50 (3.162
45	84.45 (4.140)	71.33 (1.414)	-	122.87 (6.450)	129.75 (10.214)	-	125.25 (3.426)	103.75 (2.986)	-
60	89.75 (9.899)	69.87 (10.150)	40.00 (7.549)	130.60 (8.655)	143.20 (7.000)	150.00 (0.547)	139.90 (7.127)	118.00 (6.041)	136.70 (3.774
90	84.50 (7.751)	57.87 (4.949)	35.40 (5.338)	136.70 (4.893)	135.90 (10.655)	126.00 (6.574)	133.60 (8.294)	117.00 (8.796)	105.80 (1.181
120	82.83*	55.97** (3.618)	15.40* (4.098)	135.54° (6.641)	138.10° (3.865)	61.80** (3.053)	134.42° (3.824)	118.75* (6.694)	57.80** (4.220

p<0.001



^{**} p>0.001

Moreover, at RT/Amb RH, the extent of reduction the same in strength of tablets was for formulations. Generally, direct-compressed tablets were much susceptible to moisture.

Disintegration Time

Table 5 represents the disintegration tablets exposed to 75% RH. At this condition, the remarkable decrease was found with direct-compressed tablets but no change was observed in tablets prepared with EC. The same results were also obtained at RH (Table 5).

RH, moisture sorption shortened disintegration time of tablets prepared by direct-compression and by PVP, but no change was in tablets made by EC. The extent of decrease disintegration time of tablets prepared with less than direct compressed formulations. These changes were attributed to the dissolution of solid within the tablet structure.

general, moisture sorption during storage did not affect the disintegration tablets prepared by EC. When compared two tablets granulated with PVP were much susceptible change caused by moisture than tablets granulated EC. This data indicated accordance with in report (3).

When exposing to low humidity (30% noticeable change depending on moisture loss of tablets was found (Table 5).



Table 5 Effect of Moisture on the Disintegration time (min \pm SD) of Tablets (n=5)

Time						7.	RH							
(days)		F-	·Ī			F-	-II			F-:	III			
	30	50	75	Amb	30	50	75	Amb	30	50	75	Amb		
0	5.38 (0.175)	5.38 (0.175)	5.38 (0.175)	5.38 (0.175)	13.26 (0.950)	13.93 (0.404)	13.26 (0.950)	13.26 (0.950)	12.26 (0.205)	12.21 (0.288)	12.13 (0.205)	12.13 (0.205		
i	4.45 (0.975)	4. 03 (0.879)	2.70° (0.638)	-	12.33 (0.775)	13.40 (0.983)	11.20 (0.867)	-	12.26 (1.155)	13.91 (1.282)	11.76 (0.375)	-		
2	5.36 (0.758)	3.57 (0.629)	3.36 (0.485)	-	12.61 (0.573)	12.25 (0.050)	11.65 (0.996)	-	13.07 (0.332)	13.78 (1.320)	11.80 (0.435)	-		
8	4.53 (0.680)	4.15 (1.409)	3.06 (1.001)	-	12.46 (0.144)	13.02 (0.390)	11.47 (0.153)	-	12.64 (0.314)	11.98 (0.815)	11.31 (0.076)	-		
14	4.99 (0.820)	3.96 (0.401)	2.50 (1.169)	-	12.83 (0.464)	13.48 (0.455)	11.65 (0.482)	-	12.96 (0.693)	13.13 (0.884)	11.20 (0.150)	•		
30	5.40 (0.744)		, , ,		1.95 (0.444)	3.50 (0.410)	12.65 (1.284)	12.40 (0.431)	11.11 (0.390)	12.44 (0.825)	12.41 (0.768)	12.33 (0.426)	12.14 (0.193)	14.07 (0.444
45	5.02 (0.372)	3.95 (0.261)	0.61 (0.115)	-		1 1	1 1	11.87 12.74 (0.261) (1.236)	11.96 (1.652)	-				
60		3.85 (0.900)	12.75 (0.971)	12.78 (0.794)	11.08 (0.849)	12.58 (0.660)	12.35 (0.649)	11.82 (0.823)	11.66	14.98 (0.653				
90	4.33 (0.483)	3.66 (0.978)	0.61 (0.369)	3.88 (0.595)	12.25 (0.453)	12.95 (1.162)	10.79 (0.782)	12.33 (0.545)	11.53 (1.420)	13.70 (0.512)	13.29 (0.308)	11.78 (0.707		
120	5.18° (0.270)	3.06° (0.679)	1.63**	2.31° (0.718)	12.64° (1.044)	12.11° (1.775)	11.37*	11.81° (1.121)	12.07*	12.42° (0.920)	15.32* (0.780)	11.88*		

[&]quot; p<0.001
"" p>0.001



Table 6 Effect of Moisture on the Dissolution Parameters $[k(min^-), t_{50}(min)]$ of Ranitidine hydrochloride tablets

	Tias				Fo	praulation	15			
[% RH]	Time (days)		F-I			F-II			F-111	
		k	k*	tso	k	r	tso	k	r	tso
-	0	0.0483	-0.962	14.34	0.0552	-0.988	12.55	0.0414	-0.998	16.73
***	90	0.0299	-0.887	23.17	0.0667	-0.980	10.38	0.0483	-0.972	14,34
75	120	0.0529	-0.924	13.10	0.0391	-0.980	17.72	0.0854	-0.970	8.11
	90	0.0345	-0.893	20.08	0.0575	-0.985	12.05	0.0483	-0.930	14.34
50	120	0.0460	-0.907	15.06	0.0414	-0.980	16.73	0.0437	-0.940	15.85
70	90	0.0598	-0.983	11.58	0.0368	-0.922	18.83	0.0414	-0.904	16.73
30	120	0.0414	-0.861	16.73	0.0299	-0.995	23.18	0.0437	-0.992	15.85
A-L DII	90	0.0391	-0.909	17.72	0.0483	-0.983	14.34	0.0460	-0.951	15.06
Amb RH	120	0.0437	-0.916	15.58	0.0322	-0.991	21.52	0.0529	-0.972	13.10

^{*} Correlation Coefficients

Dissolution Studies

Table 6 indicates the effect of moisture on the ranitidine dissolution parameters of hydrochloride tablets. After 120 days, a remarkable decrease observed in the dissolution rate of drug from made by PVP. This may be due to the high binding effect of PVP. On the other hand, except exposed 75% drug release from tablets prepared with EC did not



Table 7

Drug Content (%) of Ranitidine hydrochloride
 Tablets during the Storage
 (n=3)

j.						7	Z RH					
(days)		I				i.	F-11			F-111	Ш	
	88	20	7.5	Amb	30	50	75	Amb	30	50	7.5	Amb
0	97.96•	97.96 (2.13)	97.96 (2.13)	97.96 (2.13)	95.54 (1.07)	95.54 (1.07)	95.54 (1.07)	95.54 (1.07)	95.21 (7.23)	95.21 (7.23)	95.21 (7.23)	95.21 (7.23)
30	95.57 (4.86)	95.22 (5.69)	94.32 (2.53)	96.00 (4.38)	97.01	97.56 (1.47)	94.55 (0.57)	94.36 (5.05)	96.54 (3.85)	95,59	94.64 (2.26)	92,76 (6.97)
09	95.36 (1.13)	96.40 (5.20)	93.58 (4.40)	91.13	93.38 (2.88)	94.42 (2.74)	95.73 (1.65)	96.34 (3.56)	94.04 (3.95)	92,59 (1,80)	93.09	106.30 (0.84)
90	96.41 (2.67)	93.71 (3.88)	95.42 (3.79)	93.48 (5.51)	94.33 (5.70)	93.56 (4. 28)	93.00 (0.26)	93.48 (0.57)	95.76 (5.55)	93.40 (3.91)	94.37 (1.37)	95,11 (6,43)
120	93.44 (3.69)	96.71	96.34 (1.73)	92.16	96.67 (5.20)	95.44 (2.32)	93.41 (3.64)	93.24 (2.21)	97.93 (2.72)	96.43 (2.75)	94.66 (1.13)	94.18 (2.53)

* ± SD was given in the brankets.



alter. However, at 75% RH, drug release increased within the storage period.

Furthermore, as seen in Table 7 ranitidine is reasonably stable hydrochloride at the conditions.

a conclusion, tabletting technique type have an importance on physical stability of ranitidine hydrochloride tablets.

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